

HDROZONE™

HYDRONIUM-BASED HYBRID STERILIZATION

Scientific Definition of Damage-Free Sterilization

Limits in Flexible Endoscopes

Hasan Tahsin ÖZBEK • Bülent DEVECİ • Yüksel ERGÜN

White Paper - **18.08.2025**

PURPOSE

This document has been prepared to scientifically define the boundary between microbial efficacy and material integrity in the low-temperature sterilization of flexible endoscopes and to demonstrate the position of the HDROZONE™ (H₃O⁺-based hybrid sterilization) approach within this boundary.



KEYS

Endoscope, Sterilization, Hydronium, H₂O₂, Contamination, Biofilm, Risk, Validation

1. PROBLEM STATEMENT

The fundamental problem in flexible endoscope sterilization is as follows:

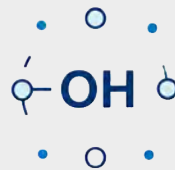
The high chemical/reactive load applied to achieve sterility in challenging lumens causes irreversible damage to the polymer, elastomer, and adhesive components of the device

This problem creates three critical contradictions:



Penetration vs. Damage

- Aggressive chemistry is required for intra-lumen sterilization.
- The same chemistry damages the outer body.



Radical Effect vs. Material Stability

- High reactive species (especially radicals) are effective in microbial killing.
- However, they break polymer chains.



Low Temperature vs. Chemical Load

- Low temperature is advantageous for the material.
- However, as the chemical load increases, this advantage disappears.

CONCLUSION:



Most current sterilization approaches either guarantee sterility but reduce device lifespan or protect the device but create risks in lumen sterility.

2. ENGINEERING FRAMEWORK FOR THE DEFINITION OF “DAMAGE-FREE” IN ENDOSCOPE STERILIZATION

For an endoscope, damage-free status must be preserved simultaneously across the following four axes:



Mechanical Integrity

- No cracks, hardening, tackiness, or micro-tears in the outer sheath.



Functional Integrity

- No change in bending, torque transmission, channel patency, or leak test results.



Optical / Electrical Integrity

- No degradation in image and light transmission, distal tip transparency, or insulation.



Chemical Safety

- Must not leave unacceptable residues or degradation products.

This approach is compatible with:

- FDA sterilization guidelines
- AAMI TIR17 material compatibility approach

3. RISK DISTRIBUTION IN ENDOSCOPE STRUCTURE

► Critical Sensitive Areas

- TPU / silicone / Pebax outer sheath
- Epoxy and adhesive regions
- Distal tip seal areas
- Elastomeric strain-relief zones

► Relatively Durable Areas

- Stainless steel spiral structure
- PTFE / FEP / ETFE channels
- Optical glass surfaces

CONCLUSION:

Sterilization must be optimized not for a single “device” but for a multi-material system.

4. ENDOSCOPE DAMAGE MODEL FOR HDROZONE™

HDROZONE generates a hybrid chemical environment. Damage arises from the following total load:

TOTAL DAMAGE LOAD ≈

Oxidative Load + Ionic Load + Moisture Effect + Temperature Effect + Time/Cycle Effect

This model is reduced to 5 control axes:

1. VH_2O_2 equivalent dose
2. O_3 equivalent dose
3. Plasma / radical fraction
4. Relative humidity / water activity
5. Number of cycles (cumulative aging)

Critical Point:

Sterilization is not a single-cycle issue but a cumulative aging problem.

5. DAMAGE-FREE OPERATION WINDOW

REGION A — SAFE WINDOW

- Heavy oxidant dominant
 - Low radicals
 - Low temperature
- Controlled humidity
 - Low residue

RESULT:

- Microbial killing is achieved
- Polymer damage is minima



HDROZONE target regime

REGION B — EFFECTIVE BUT STRESSFUL

- More aggressive chemistry

RISK:

- Hardening
- Color Change
- Seal Fatigue
- Adhesive Damage



**Only acceptable with
validation**

REGION C — DAMAGE ZONE

- High radical density
 - Long exposure
 - High humidity

RESULT:

- Polymer Swelling
 - Oxidation
- Deformation



Must be strictly avoided

6. SCIENTIFIC DESIGN RULES

► Rule 1 — Heavy Oxidant Dominance, Not Radicals

- High radicals - chain scission

SOLUTION:

- Keep OH· low
- Plasma should be an auxiliary phase

► Rule 2 — Optimal Humidity

- Low humidity - poor transport
- High humidity - aggressive chemistry

TARGET:

- “Humidity present but no wet surface”

► Rule 3 — Stable Temperature

- Low temperature is an advantage
- However, local temperature peaks are risky

► Rule 4 — The Outer Surface Must Not Be Sacrificed for the Lumen

This is critical:

Sterilization optimization is not only about SAL.

New Metric:

- Lumen sterility / outer surface damage ratio

7. VALIDATION THRESHOLD ARCHITECTURE

Primary Criteria

- Stable leak test
- No change in bending
- Unchanged channel flow
- Preserved optical quality
 - No surface damage

Secondary Criteria

- Limited hardness change
 - Preserved flexibility
- Preserved adhesive strength
- Limited oxidation index
 - Low ΔE color change
 - Residue below limit

Red Criteria (Reject)

- Distal haze / fogging
 - Stickiness
 - Micro-cracks
- Bending deviation
- Whitening / crazing
- Increased lumen friction

8. TESTING AND VERIFICATION PLAN ON ENDOSCOPES

► Phase 1 — Base Materials Used in Endoscopes:

- TPU, silicone, Pebax
- PTFE / FEP
- Epoxy
- Lens
- Elastomers

TEST:

- 1x – 100x cycles
- Oxidative load variation
- Humidity variation
- Plasma on/off

MEASUREMENTS:

- FTIR
- SEM
- Contact angle
- Hardness
- Tensile
- Mass change
- Color



► Phase 2 — Sub-systems

- Insertion tube
- Bending section
- Channel
- Distal tip

► Phase 3 — Complete Device

- Leak test
- Articulation
- Imaging
- Flow test
- Residue
- BI/CI

9. SCIENTIFIC CONCLUSION

1. Damage-free boundary ≠ maximum microbial inactivation
2. Damage-free boundary = lumen sterility + material preservation
3. HDROZONE Advantage:

It offers higher material compatibility potential compared to classical plasma systems in a heavy oxidant-dominant – low radical regime.

The future of flexible endoscope sterilization:

It is not more aggressive chemistry, but controlled chemical engineering.

The HDROZONE approach constitutes the engineering foundation of this transition.

REFERENCES

Bu Daniela Trogolo, J. Samuel Arey, Peter R. Tentscher. Gas-Phase Ozone Reactions with a Structurally Diverse Set of Molecules: Barrier Heights and Reaction Energies Evaluated by Coupled Cluster and Density Functional Theory Calculations. *The Journal of Physical Chemistry A* 2019, 123 (2) , 517-536. DOI: 10.1021/acs.jpca.8b10323.

1. Alexandra Fischbacher, Katja Löppenber, Clemens von Sonntag, and Torsten C. Schmidt . A New Reaction Pathway for Bromite to Bromate in the Ozonation of Bromide. *Environmental Science & Technology* 2015, 49 (19) , 11714-11720. DOI: 10.1021/acs.est.5b02634.

2. Yi Yang, Jin Jiang, Jinglin Lu, Jun Ma, and Yongze Liu . Production of Sulfate Radical and Hydroxyl Radical by Reaction of Ozone with Peroxymonosulfate: A Novel Advanced Oxidation Process. *Environmental Science & Technology* 2015, 49 (12) , 7330-7339. DOI: 10.1021/es506362e.

3. Janez Cerkovnik and Božo Plesničar . Recent Advances in the Chemistry of Hydrogen Trioxide (HOOH). *Chemical Reviews* 2013, 113 (10) , 7930-7951. DOI: 10.1021/cr300512s.

4. Alexandra Fischbacher, Justus von Sonntag, Clemens von Sonntag, and Torsten C. Schmidt . The •OH Radical Yield in the H₂O₂ + O₃ (Peroxone) Reaction. *Environmental Science & Technology* 2013, 47 (17) , 9959-9964. DOI: 10.1021/es402305r.

5. Carlos Barrera-Díaz, Lina. A. Bernal-Martínez, Reyna Natividad, and Juan M. Peralta-Hernández . Synergy of Electrochemical/O₃ Process with Aluminum Electrodes in Industrial Wastewater Treatment. *Industrial & Engineering Chemistry Research* 2012, 51 (27) , 9335-9342. DOI: 10.1021/ie3004144.

6. Sergej Naumov and Clemens von Sonntag . Standard Gibbs Free Energies of Reactions of Ozone with Free Radicals in Aqueous Solution: Quantum-Chemical Calculations. *Environmental Science & Technology* 2011, 45 (21) , 9195-9204. DOI: 10.1021/es2018658.

7. J. Pablo Pocostales, Myint M. Sein, Wolfgang Knolle, Clemens von Sonntag, and Torsten C. Schmidt. Degradation of Ozone-Refractory Organic Phosphates in Wastewater by Ozone and Ozone/Hydrogen Peroxide (Peroxone): The Role of Ozone Consumption by Dissolved Organic Matter. *Environmental Science & Technology* 2010, 44 (21) , 8248-8253. DOI: 10.1021/es1018288.

8. Helena Jablonowski, Joao Santos Sousa, Klaus-Dieter Weltmann, Kristian Wende, Stephan Reuter. Quantification of the ozone and singlet delta oxygen produced in gas and liquid phases by a non-thermal atmospheric plasma with relevance for medical treatment. *Scientific Reports* 2018, 8 (1) DOI: 10.1038/s41598-018-30483-w.

9. Yalei Ding, Jiejie Wang, Shanshan. u, Kun-Yi Andrew Lin, Shaoping Tong. Oxygen vacancy of CeO₂ improved efficiency of H₂O₂/O₃ for the degradation of acetic acid in acidic solutions. *Separation and Purification Technology* 2018, 207, 92-98. DOI: 10.1016/j.seppur.2018.06.027.

10. Donggwan Lee, Jae-Cheol Lee, Joo-Youn Nam, Hyun-Woo Kim. Degradation of sulfonamide antibiotics and their intermediates toxicity in an aeration-assisted non-thermal plasma while treating strong wastewater. *Chemosphere* 2018, 209, 901-907. DOI: 10.1016/j.chemosphere.2018.06.125.
11. David B. Miklos, Christian Remy, Martin Jekel, Karl G. Linden, Jörg E. Drewes, Uwe Hübner. Evaluation of advanced oxidation processes for water and wastewater treatment – A critical review. *Water Research* 2018, 139, 118-131. DOI: 10.1016/j.watres.2018.03.042.
12. Pierre-François Biard, Thom Thi Dang, Jenny Bocanegra, Annabelle Couvert. Intensification of the O₃/H₂O₂ advanced oxidation process using a continuous tubular reactor filled with static mixers: Proof of concept. *Chemical Engineering Journal* 2018, 344, 574-582. DOI: 10.1016/j.cej.2018.03.112.
13. Emmanuel Mousset, Nihal Oturan, Mehmet A. Oturan. An unprecedented route of OH radical reactivity evidenced by an electrocatalytical process: Ipso-substitution with perhalogenocarbon compounds. *Applied Catalysis B: Environmental* 2018, 226, 135-146. DOI: 10.1016/j.apcatb.2017.12.028.
14. J.F. Pérez, S. Sabatino, A. Galia, M.A. Rodrigo, J. Llanos, C. Sáez, O. Scialdone. Effect of air pressure on the electro-Fenton process at carbon felt electrodes. *Electrochimica Acta* 2018, 273, 447-453. DOI: 10.1016/j.electacta.2018.04.031.
15. Guilherme Garcia Bessegato, João Carlos de Souza, Juliano Carvalho Cardoso, Maria Valnice Boldrin Zanoni. Assessment of several advanced oxidation processes applied in the treatment of environmental concern constituents from a real hair dye wastewater. *Journal of Environmental Chemical Engineering* 2018, 6 (2) , 2794-2802. DOI: 10.1016/j.jece.2018.04.041.
16. A. Privat-Maldonado, Y. Gorbanev, D. O'Connell, R. Vann, V. Chechik, M. W. van der Woude. Nontarget Biomolecules Alter Macromolecular Changes Induced by Bactericidal Low-Temperature Plasma. *IEEE Transactions on Radiation and Plasma Medical Sciences* 2018, 2 (2) , 121-128. DOI: 10.1109/TRPMS.2017.2761405.
17. Mika Sillanpää, Mohamed Chaker Ncibi, Anu Matilainen. Advanced oxidation processes for the removal of natural organic matter from drinking water sources: A comprehensive review. *Journal of Environmental Management* 2018, 208, 56-76. DOI: 10.1016/j.jenvman.2017.12.009.
18. B.W. Darvell. *More Chemistry*. 2018,, 771-789. DOI: 10.1016/B978-0-08-101035-8.50030-4.
19. Erika Reisz, Clemens von Sonntag, Agnes Tekle-Röttering, Sergej Naumov, Winfried Schmidt, Torsten C. Schmidt. Reaction of 2-propanol with ozone in aqueous media. *Water Research* 2018, 128, 171-182. DOI: 10.1016/j.watres.2017.10.035.
20. João F. Gomes, Inês Leal, Katarzyna Bednarczyk, Marta Gmurek, Marek Stelmachowski, Magdalena Diak, M. Emília Quinta-Ferreira, Raquel Costa, Rosa M. Quinta-Ferreira, Rui C. Martins. Photocatalytic ozonation using doped TiO₂ catalysts for the removal of parabens in water. *Science of The Total Environment* 2017, 609, 329-340. DOI: 10.1016/j.scitotenv.2017.07.180.

21. Pierre-François Biard, Thom Thi Dang, Annabelle Couvert. Determination by reactive absorption of the rate constant of the ozone reaction with the hydroperoxide anion. *Chemical Engineering Research and Design* 2017, 127, 62-71. DOI: 10.1016/j.cherd.2017.09.004.
22. Ozge Turkay, Sibel Barışçı, Mika Sillanpää. E-peroxone process for the treatment of laundry wastewater: A case study. *Journal of Environmental Chemical Engineering* 2017, 5 (5) , 4282-4290. DOI: 10.1016/j.jece.2017.08.012.
23. Minhwan Kwon, Homin Kye, Youmi Jung, Yejoon Yoon, Joon-Wun Kang. Performance characterization and kinetic modeling of ozonation using a new method: R OH,O₃ concept. *Water Research* 2017, 122, 172-182. DOI: 10.1016/j.watres.2017.05.062.
24. Ligy Philip, Bhallamudi Murty, Channarong Puchongkawarin, Miao Guo, Nilay Shah, David Stuckey, Benoit Chachuat, Yannic Vaupel, Sarojini Tiwari, Chitta Behera, Babji Srinivasan, Chedly Tizaoui, Olajumoke Odejimi, Ayman Abdelaziz. 3 *Wastewater Treatment*. 2017,, 115-180. DOI: 10.1201/9781315153209-4.
25. Songjie Wu, Qian Zhang, Ruonan Ma, Shuang Yu, Kaile Wang, Jue Zhang, Jing Fang. Reactive radical-driven bacterial inactivation by hydrogen-peroxide-enhanced plasma-activated-water. *The European Physical Journal Special Topics* 2017, 226 (13) , 2887-2899. DOI: 10.1140/epjst/e2016-60330-y.
26. Julia Patzsch, Jonathan Z. Bloh. Improved photocatalytic ozone abatement over transition metal-grafted titanium dioxide. *Catalysis Today* 2017, DOI: 10.1016/j.cattod.2017.07.010.
27. Xinyang Li, Shaobin Sun., u Zhang, Guicheng Liu, Clark Renjun Zheng, Jianzhong Zheng, Dayi Zhang, Hong Yao. Combined electro-catazone/electro-peroxone process for rapid and effective Rhodamine B degradation. *Separation and Purification Technology* 2017, 178, 189-192. DOI: 10.1016/j.seppur.2016.12.052.
28. Jingxin Yang, Ji Li, Wenyi Dong, Jun Ma, Jiayin Li. Influence of nitrite on the degradation of atrazine by ozonation. *Journal of Chemical Technology & Biotechnology* 2017, 92 (2) , 442-450. DOI: 10.1002/jctb.5031.
29. George Wafula Wanjala, Arnold Onyango, Calvin Onyango, Moses Makayoto. . *African Journal of Biochemistry Research* 2017,, 79. DOI: 10.5897/AJBR2017.0967.
30. Glen Andrew de Vera, Wolfgang Gernjak, Howard Weinberg, Maria José Farré, Jurg Keller, Urs von Gunten. Kinetics and mechanisms of nitrate and ammonium formation during ozonation of dissolved organic nitrogen. *Water Research* 2017, 108, 451-461. DOI: 10.1016/j.watres.2016.10.021.
31. M. E. Zappi, R. Hernandez, D. Gang, R. Bajpai, C. H. Kuo, D. O. Hill. Treatment of groundwater contaminated with high levels of explosives using advanced oxidation processes. *International Journal of Environmental Science and Technology* 2016, 13 (12) , 2767-2778. DOI: 10.1007/s13762-016-1109-x.
32. Juhong Zhan, Yujue Wang, Huijiao Wang, Wenhua Shen., uejun Pan, Jinlin Wang, Gang Yu. Electro-peroxone regeneration of phenol-saturated activated carbon fiber: The effects of irreversible adsorption and operational parameters. *Carbon* 2016, 109, 321-330. DOI: 10.1016/j.carbon.2016.08.034.

33. Jingxin Yang, Ji Li, Wenyi Dong, Jun Ma, Jie Cao, Tingting Li, Jiayin Li, Jia Gu, Pingxin Liu. Study on enhanced degradation of atrazine by ozonation in the presence of hydroxylamine. *Journal of Hazardous Materials* 2016, 316, 110-121. DOI: 10.1016/j.jhazmat.2016.04.078.
34. Xin Cheng, Hongguang Guo, Hongwei Liu, Yang Liu, Ying Yang, Yongli Zhang. Performance and Mechanism on Degradation of Estriol Using O₃ /PS Process. *Ozone: Science & Engineering* 2016, 38 (5) , 358-366. DOI: 10.1080/01919512.2016.1170589.
35. László Wojnárovits, Erzsébet Takács. Radiation Induced Degradation of Organic Pollutants in Waters and Wastewaters. *Topics in Current Chemistry* 2016, 374 (4) DOI: 10.1007/s41061-016-0050-2.
36. Jaedon Shin, Zahra Ramadhany Hidayat, Yunho Lee. Influence of Seasonal Variation of Water Temperature and Dissolved Organic Matter on Ozone and OH Radical Reaction Kinetics During Ozonation of a Lake Water. *Ozone: Science & Engineering* 2016, 38, 100-114. DOI: 10.1080/01919512.2015.1079120.
37. Yury Gorbanev, Deborah O'Connell, Victor Chechik. Non-Thermal Plasma in Contact with Water: The Origin of Species. *Chemistry - A European Journal* 2016, 22 (10) , 3496-3505. DOI: 10.1002/chem.201503771.
38. Arnold N. Onyango. Alternatives to the 'water oxidation pathway' of biological ozone formation. *Journal of Chemical Biology* 2016, 9, 1-8. DOI: 10.1007/s12154-015-0140-6.
39. Fei Qi, Wei Chu, Bingbing. u. Comparison of phenacetin degradation in aqueous solutions by catalytic ozonation with CuFe₂O₄ and its precursor: Surface properties, intermediates and reaction mechanisms. *Chemical Engineering Journal* 2016, 284, 28-36. DOI: 10.1016/j.cej.2015.07.095.
40. Erika Reisz, Sergej Naumov, Winfried Schmidt, Clemens von Sonntag. Reaction of Ozone with Ag(I)—Mechanistic Considerations. *Ozone: Science & Engineering* 2015, 37, 393-404. DOI: 10.1080/01919512.2015.1041583.
41. Huijiao Wang, Shi Yuan, Juhong Zhan, Yujue Wang, Gang Yu, Shubo Deng, Jun Huang, Bin Wang. Mechanisms of enhanced total organic carbon elimination from oxalic acid solutions by electro-peroxone process. *Water Research* 2015, 80, 20-29. DOI: 10.1016/j.watres.2015.05.024.
42. Jesús Ferre-Aracil, Salvador C. Cardona, Javier Navarro-Laboulais. Kinetic Study of Ozone Decay in Homogeneous Phosphate-Buffered Medium. *Ozone: Science & Engineering* 2015, 37, 330-342. DOI: 10.1080/01919512.2014.998756.
43. Fei Qi, Bingbing. u, Wei Chu. Heterogeneous catalytic ozonation of phenacetin in water using magnetic spinel ferrite as catalyst: Comparison of surface property and efficiency. *Journal of Molecular Catalysis A: Chemical* 2015, 396, 164-173. DOI: 10.1016/j.molcata.2014.10.001.
44. Erika Reisz, Alexandra Fischbacher, Sergej Naumov, Clemens von Sonntag, Torsten C. Schmidt. Hydride Transfer: A Dominating Reaction of Ozone with Tertiary Butanol and Formate Ion in Aqueous Solution. *Ozone: Science & Engineering* 2014, 36, 532-539. DOI: 10.1080/01919512.2014.891436.
45. Sergey L. Khursan. Peroxide intermediates of oxidation processes: Organic trioxides. 2014,,, 1-72. DOI: 10.1002/9780470682531.pat0874.

46. S. Reed Plimpton, Mark Gołkowski, Deborah G. Mitchell, Chad Austin, Sandra S. Eaton, Gareth R. Eaton, Czesław Gołkowski, Martin Voskuil. Remote delivery of hydroxyl radicals via secondary chemistry of a nonthermal plasma effluent. *Biotechnology and Bioengineering* 2013, 110 (10.1002/bit.v110.7) , 1936-1944. DOI: 10.1002/bit.24853.
47. Zdenko Machala, Barbora Tarabova, Karol Hensel, Eva Spetlikova, Libusa Sikurova, Petr Lukes. Formation of ROS and RNS in Water Electro-Sprayed through Transient Spark Discharge in Air and their Bactericidal Effects. *Plasma Processes and Polymers* 2013, 10 (10.1002/ppap.v10.7) , 649-659. DOI: 10.1002/ppap.201200113.
48. Josep M. Anglada, Miquel Torrent-Sucarrat, Manuel F. Ruiz-Lopez, Marilia Martins-Costa. Is the HO₄⁻ Anion a Key Species in the Aqueous-Phase Decomposition of Ozone?. *Chemistry - A European Journal* 2012, 18 (10.1002/chem.v18.42) , 13435-13445. DOI: 10.1002/chem.201201991.
49. Marek Golkowski, Czesław Golkowski, Jori Leszczynski, S. Reed Plimpton, Piotr Maslowski, Aleksandra Foltynowicz, Jun Ye, Bruce McCollister. Hydrogen-Peroxide-Enhanced Nonthermal Plasma Effluent for Biomedical Applications. *IEEE Transactions on Plasma Science* 2012, 40, 1984-1991. DOI: 10.1109/TPS.2012.2200910.
50. D. Gardoni, A. Vailati, R. Canziani. Decay of Ozone in Water: A Review. *Ozone: Science & Engineering* 2012, 34, 233-242. DOI: 10.1080/01919512.2012.686354.
51. Sergej Naumov, Clemens von Sonntag. The reaction of •OH with O₂, the decay of O₃^{•-} and the pKa of HO₃^{•-} - interrelated questions in aqueous free-radical chemistry. *Journal of Physical Organic Chemistry* 2011, 24 (10.1002/poc.v24.7) , 600-602. DOI: 10.1002/poc.1812.
52. Rossano Amadelli, Luca Samiolo, Achille De Battisti, Alexander B. Velichenko. Electro-oxidation of Some Phenolic Compounds by Electrogenerated O₃ and by Direct Electrolysis at PbO₂ Anodes. *Journal of The Electrochemical Society* 2011, 158, P87. DOI: 10.1149/1.3589913.
53. Pentax, White Papers
54. ASP, White Papers
55. Olympus, White Papers